

Evaluation of the long-term behaviour of potential plutonium wasteforms in a geological repository

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Background

- Plutonium (Pu)
 - forms in reactors using U-based fuels by various nuclear reactions (mainly n, γ)
 - separated during spent fuel reprocessing (e.g. PUREX process) and stored mainly as calcined PuO_2
 - isotopic composition variable depending on reactor type and fuel burn-up (predominantly Pu-239 and Pu-240)

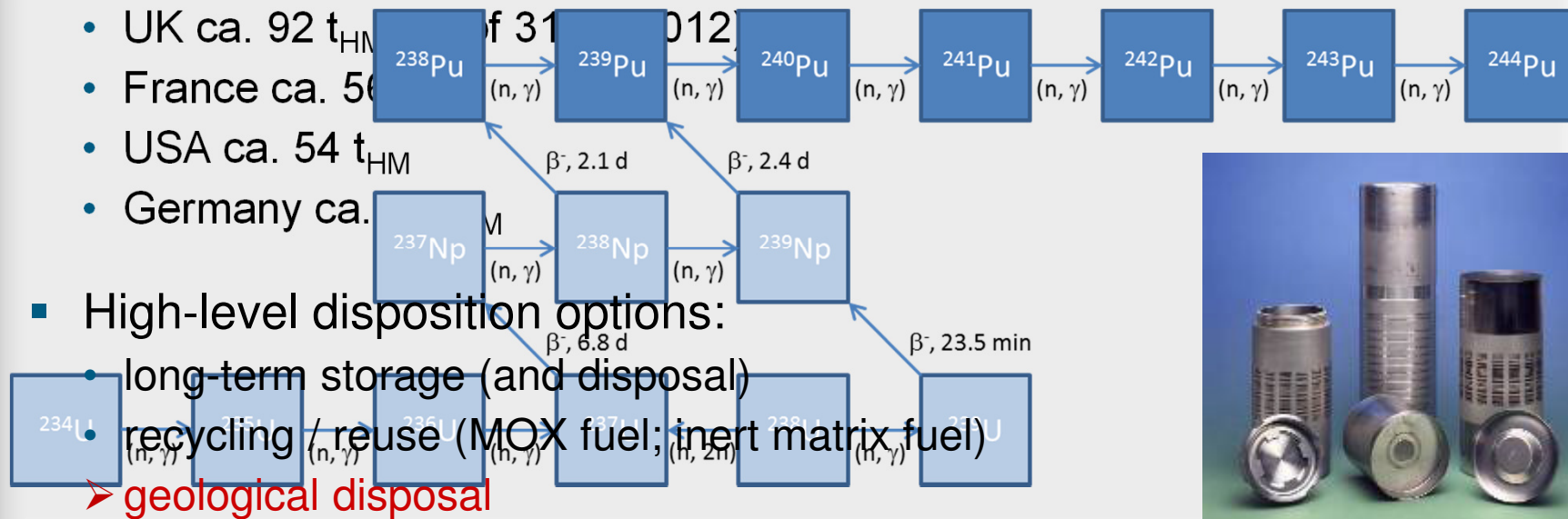
Stocks of separated civilian plutonium (2010)

- UK ca. 92 t_{HM} of 31 012
- France ca. 56 t_{HM}
- USA ca. 54 t_{HM}
- Germany ca. 1 t_{HM}

High-level disposition options:

- long-term storage (and disposal)
- recycling / reuse (MOX fuel; inert matrix fuel)

➤ geological disposal



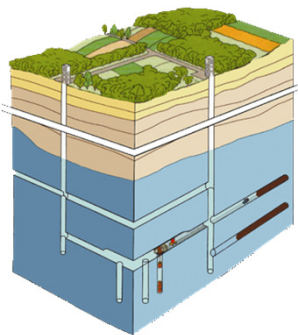
Wasteform issues for Pu disposal



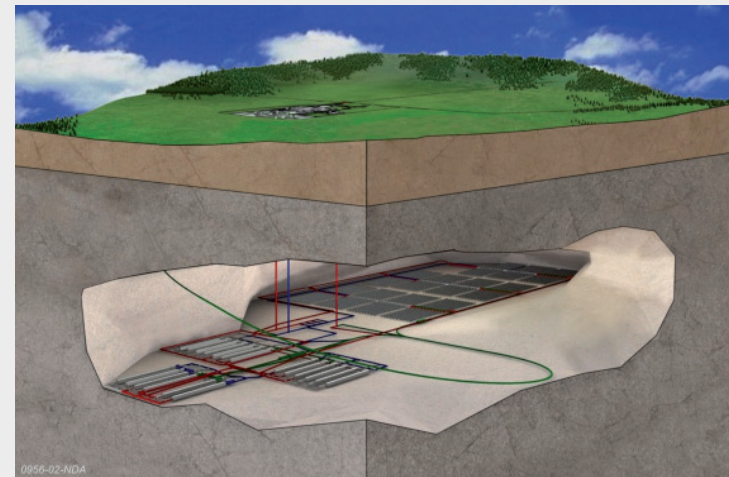
Conversion



Geological
Disposal



- Safety case issues for **geological disposal**
 - long-term **durability** of Pu wasteforms
 - long-term **radionuclide release** from Pu wasteforms under disposal conditions
 - post closure **criticality** events



Scope & Objectives

- NDA RWMD responsible for the implementation of geological disposal of higher activity wastes in the UK
- wastes to be managed through geological disposal:
 - HLW, ILW, LLW (if unsuitable for LLWR)
 - and potentially
 - spent fuel (SF)
 - **separated civil plutonium**
 - uranium (DNLEU)if declared as wastes



- Objectives:
 - Review and evaluation of performance and long-term behaviour of potential Pu wasteforms under repository conditions relevant for the UK
 - Corrosion rates / Pu release under disposal conditions
 - support decisions on Pu disposal in the UK

Geological disposal in the UK

- UK geological disposal programme in generic stage
 - generic host rock types
 - illustrative disposal concepts (incl. co-disposal of LILW)

Parameter	"higher strength" crystalline rocks	"lower strength" sedimentary rocks	Evaporites
pH	near neutral	near neutral	~7.0
E _H [mV]	< -200	< -200	< -200
dominant cations	Na	Na, Ca	Na (Ca)
Cl ⁻ [mg L ⁻¹]	13,000 ... 104,000	5,600 ... 104,000	~200,000
SO ₄ ⁻² [mg L ⁻¹]	1,000 ... 3,400	2,300 ... 3,400	3,000
HCO ₃ ⁻ [mg L ⁻¹]	< 100	100 ... 200	50
Watson et al. 2007, King 2009			

Potential Pu wasteforms

Glasses

Glass ceramics

Ceramics

Storage MOX

Encapsulants

- Borosilicate glasses
 - lanthanide borosilicate
 - lead borosilicate
 - calcium borosilicate



- Alkali-Tin-Silicate glasses

- Phosphate glasses
 - iron phosphate
 - aluminium phosphate

- Low-specification (“storage”) MOX
 - MOX fuel not destined for reactor usage (fabricated by established technology)
 - reduced technological specifications
 - higher Pu load compared



- Cements

- Polymers



■ Focal points:

- durability in aqueous environments
- wasteform stability/integrity over geological time scales (effects of radiation damage, helium build-up)
- criticality control

? Sufficient information available for the assessment of the long-term behaviour and performance of the wasteforms ?

■ Key controls of wasteform performance:

➤ intrinsic (wasteform)

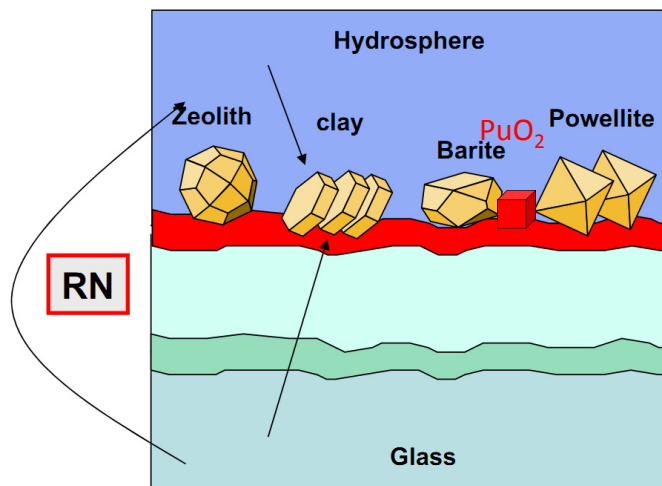
- plutonium loading
- mode of Pu incorporation
- chemical/mineralogical composition
- radiation tolerance
- mechanical stability

➤ extrinsic (repository environment)

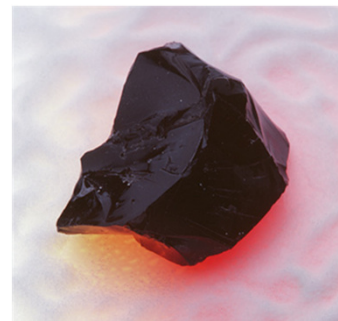
- hydrogeology (advective vs. diffusive flows)
- geochemical conditions (pH, E_H , T, I, A^+ , B^- , ...)
- microbial activity
- dependent on repository design (EBS) and host rock

Glass wasteforms for Pu

- comparatively few studies on the durability of Pu waste glasses compared to HLW glasses
- focus mainly on borosilicate / LnBS glasses
- leaching experiments mainly with standardised tests (MCC, PCT) using deionised water, mainly with surrogates (Ce, Hf), few long-term tests
- leaching rates for Pu (or surrogates) significantly lower than for glass matrix elements (e.g. B, Si, etc.)
- Pu (and surrogates) retained in secondary phases
- no effect of radiation dose on Pu release observed for borosilicate glass



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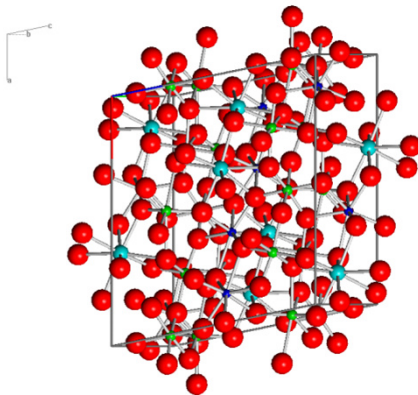
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Ceramic wasteforms for Pu

- investigations performed on various polyphase and singlephase ceramics for actinide/Pu immobilisation
- leaching experiments often with standardised tests (MCC, PCT) using deionised water or under acidic conditions, often with surrogates (Ce, Hf, Nd), long-term tests rare
 - leaching rates are orders of magnitude lower compared to glasses with rates for Pu (and surrogates), Ti and Zr often $\sim 10^{-5} \text{ g m}^{-2} \text{ d}^{-1}$ or less
 - different response to self-irradiation
 - no amorphisation of monazite and Zr-pyrochlore
 - no effect of radiation dose on durability observed for some metamict zirconolites



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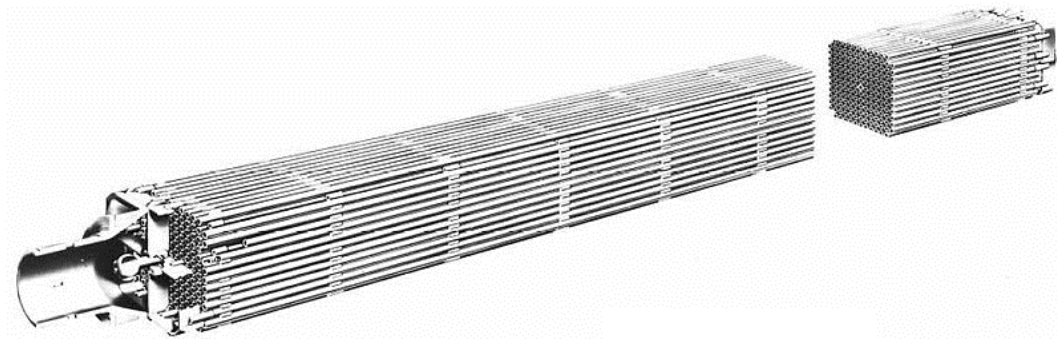
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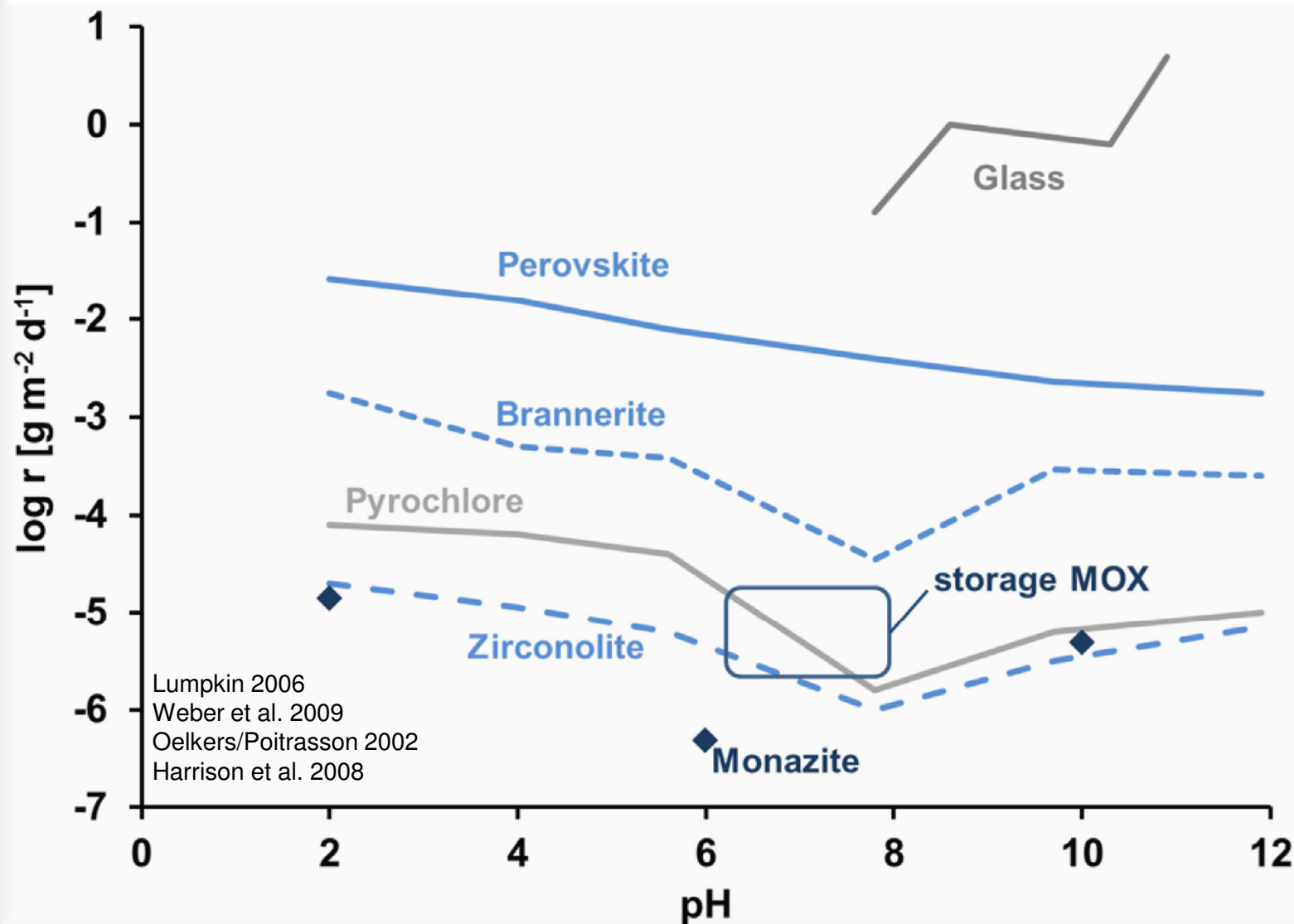
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Storage MOX as a wasteform

- concept has received considerably less attention than immobilisation of plutonium in glasses and/or ceramics
- very limited information on wasteform behaviour available
- short-term static leaching tests in deionised water, granitic water and carbonated water suggest leaching rates around $10^{-5} \text{ g m}^{-2} \text{ d}^{-1}$
- Pu leaching rates under reducing conditions lower than for uranium
- high durability under reducing (long-term) conditions in a GDF inferred



Key findings



- Assessment of corrosion rates of Pu wasteforms in a GDF?
 - understanding of wasteform behaviour
 - information for PCSA
 - safety margins of highly durable matrices
 - potential for post-closure criticality events (leaching of neutron absorbers)
 - evaluation of disturbed repository evolution scenarios (human intrusion, tectonic events/fracturing)

↪ Derivation of “bounding values” for corrosion rates under UK disposal conditions based on

- experimental data from Pu wasteforms
- analogue evidence (e.g. HLW-glasses, spent UOX/MOX)

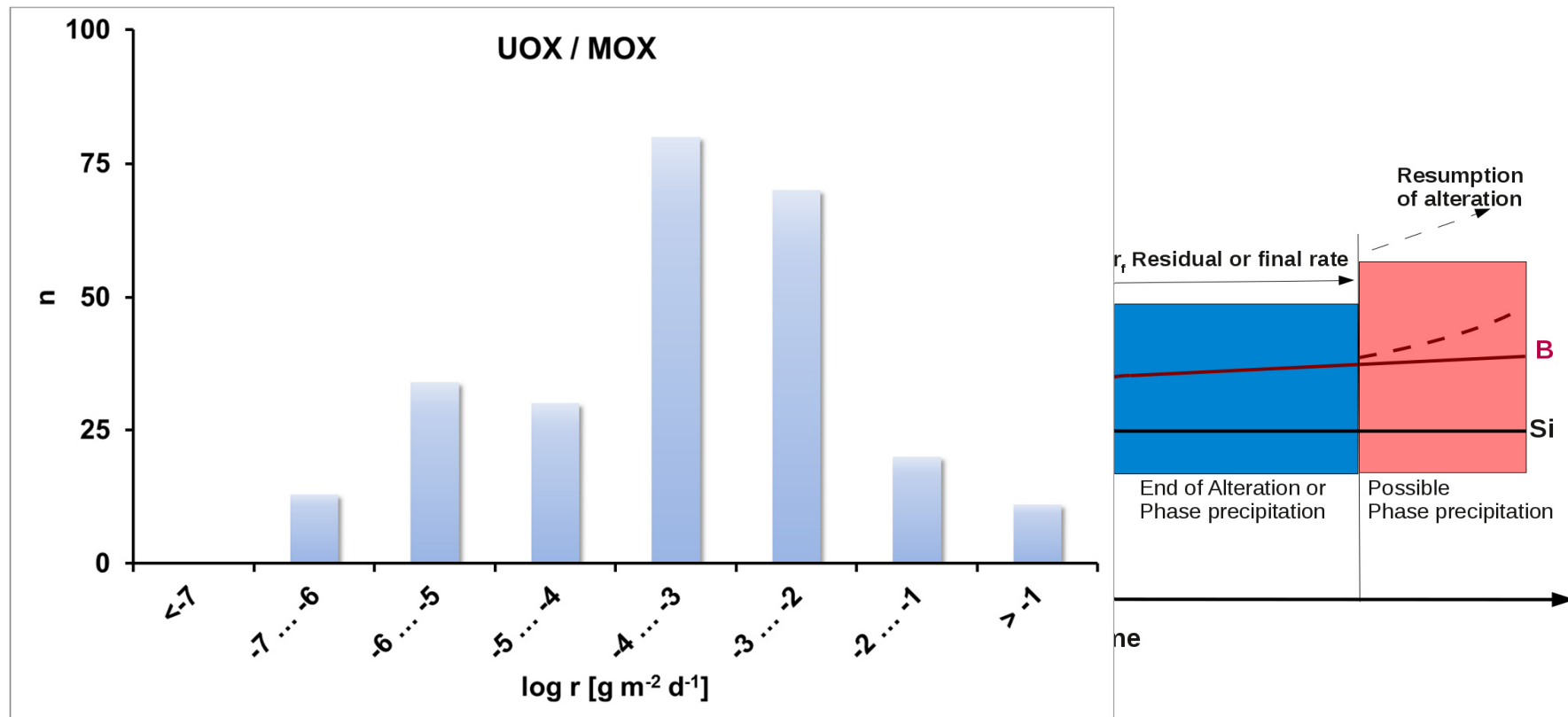
➤ Generic wasteform types

- “glasses”
- “ceramics”
- “storage MOX”

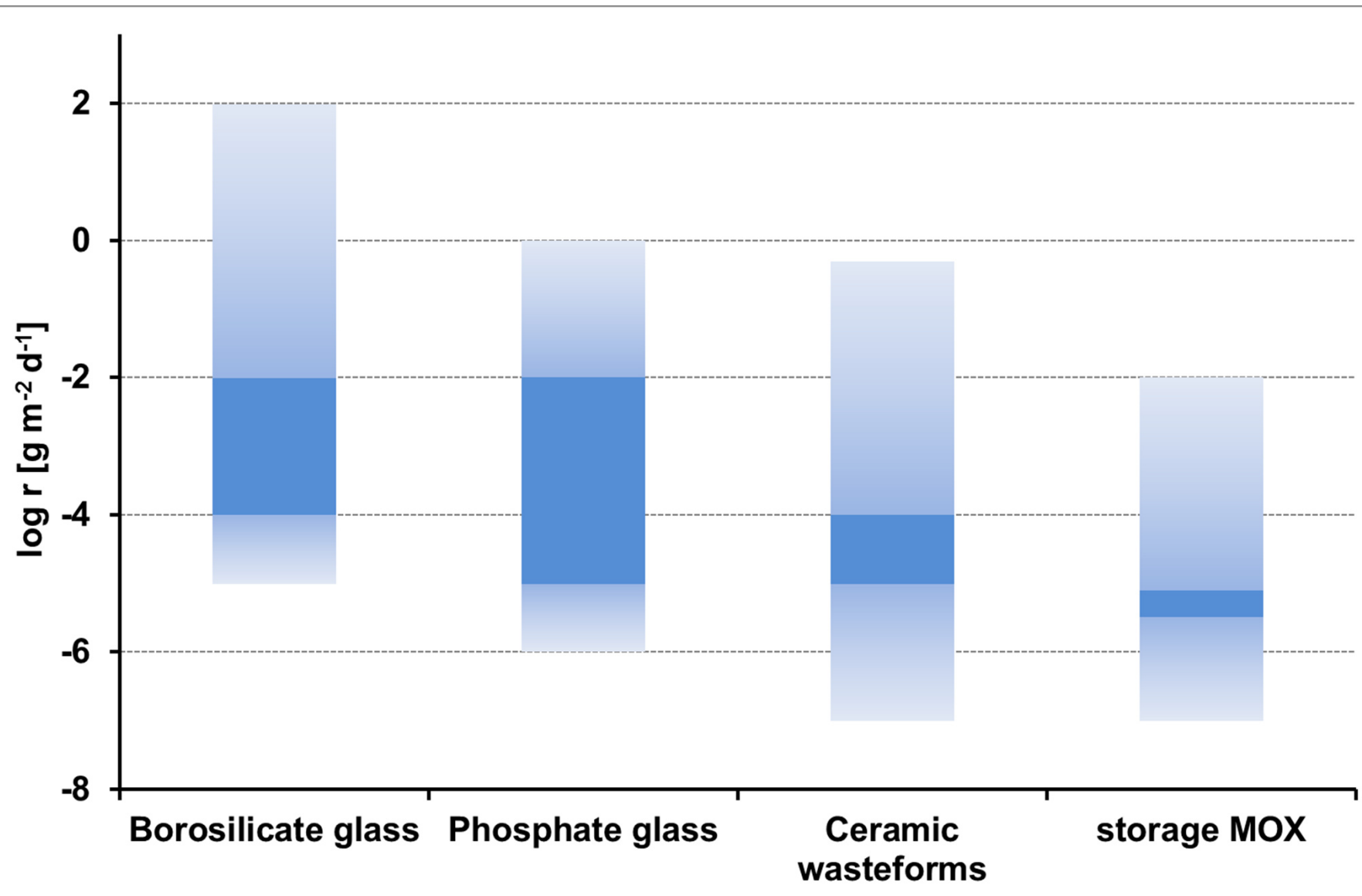
➤ near-field conditions: pH 7 ... 13; low/high salinity, ...

Derivation of corrosion rates

- storage MOX:
 - analogue SF/MOX matrix alteration (e.g. SFS, NF-PRO, MICADO)
 - conservative “upper bound”: oxidative/radiolytic SF dissolution
 - non-conservative “lower bound”: long-term SF dissolution rates (H_2 overpressure)
 - best estimate: SF dissolution rates, reducing conditions



Durability under disposal conditions



Conclusions & Outlook

- Information on the long-term behaviour and durability of plutonium wasteforms under disposal conditions rather limited compared to HLW-glasses and spent fuel
 - available information is focussed mainly on ceramics and glasses
 - ceramic wasteforms (and MOX) look promising with respect to aqueous durability
 - detailed understanding of relevant processes that govern wasteform corrosion, radionuclide release and total systems behaviour seems to be still missing
- Bounding values for plutonium wasteform corrosion rates under repository conditions can be derived from available experimental data and analogue evidence
- However:
 - more realistic assessments of wasteform durability and radionuclide release behaviour would require systematic studies regarding
 - Pu wasteform corrosion under realistic conditions
 - secondary phases formed during Pu wasteform corrosion
 - effects of self-irradiation on wasteform performance/durability
 - to explore the safety margins of the various potential wasteforms

The End

Thanks to



- NDA RWMD: for funding this project
- The Audience: for kind attention



Questions???